

## Modern Chemistry Review Stoichiometry Section 1 Answers

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### Modern Chemistry Review Stoichiometry Section

CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Given the following equation:  $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$   
4 a. What is the value of the coefficient x in this equation? 40.07 g/mol b. What is the molar mass of  $C_3H_4$ ? 2 mol O<sub>2</sub>:1 mol H<sub>2</sub>O c. What is the mole ratio ...

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Modern Chemistry 77 Stoichiometry CHAPTER 9 REVIEW Stoichiometry SECTION 3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. \_\_\_\_ The actual yield

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of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N<sub>2</sub> are mixed with 12.0 mol of H<sub>2</sub>

## CHAPTER 9 REVIEW Stoichiometry

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### Chapter 9 Review Stoichiometry Section 3

Modern Chemistry Stoichiometry Chapter 9 Section 1 Review ... CHAPTER 9 REVIEW Stoichiometry SECTION 3 PROBLEMS Write the answer on the line to the left Show all your work in the space provided 1 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g Calculate the percentage yield 2 60 mol of N<sub>2</sub> are mixed with 120 mol of H<sub>2</sub>

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## **Modern Chemistry Chapter 9 Homework 9 1 Answers**

Much of our knowledge of chemistry is based on the careful quantitative analysis of substances involved in chemical reactions. Composition stoichiometry deals with the mass relationships of elements in compounds. Reaction stoichiometry involves the mass relationships between reactants and products in a chemical reaction.

## **CorrectionKey=NL-A DO NOT EDIT--Changes must be made ...**

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