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Section 10.2 Mass and  
the Mole Section 10.3

Moles of Compounds

Section 10.4 Empirical

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biological, physical.

Name the two  
categories of science.  
the science of different

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kinds of matter and how that matter can change. Define Chemistry. organic. the study of carbon based compounds. inorganic. the study of non-organic compounds.

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States of Matter 12

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...

Elena is a chemist, a  
scientist who studies  
matter and the  
changes that it  
undergoes. Carl is very

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dependable. His teachers and his parents know that he is reliable and can be trusted. An air of jubilation surrounded the members of the science team as they received their medals for first place in the national competition.

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Physical & Chemical  
Changes - YouTubeThis  
video describes

physical and chemical  
changes in matter.

Physical Changes.

Another way to think  
about this is that a  
physical change does  
not cause a substance

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to become a fundamentally different substance but a chemical change causes a substance to change into something chemically new.

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The Mole  
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161 Section 10.1  
Measuring Matter page

320-324 Practice  
Problems pages

323-324 1. Zinc (Zn) is  
used to form a

corrosion-inhibiting  
surface on galvanized

steel. Determine the  
number of Zn atoms in

2.50 mol of Zn. 2.50

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mol Zn  
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MASTE In the space at  
the left, write true if  
the statement is true; if  
the statement is false,  
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word or phrase to  
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In this chapter you will  
learn- Concept of  
matter.Matter in our  
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chapter). Matter in our  
surroundings

Notes,Notes and

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Name Date CHAPTER  
FOR Class Section 8.2

continued 9 What is  
the relationship  
between lattice energy  
and the strength of the

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attractive force holding ions in place? a. The more positive the lattice energy is, the greater the force. b. The more negative the lattice energy is, the greater the force.

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Chemistry is central to  
all of the sciences.

CHAPTER 1 Tartaric  
*Page 24/26*



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Chapter 1 Study

Acid Crystals. MATTER  
AND CHANGE 3  
SECTION 1 OBJECTIVES  
Define chemistry. List  
examples of the  
branches of chemistry.  
Compare and contrast  
basic research, applied  
research, and  
technological  
development.

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